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Solution: Let  $a$  be the length of the sides of the tetrahedral cube and  $r$  be the radius of the two types of atoms.

$$1 a^3 = r + r^2 C H a = 2r_C + 2r_H = 2 (0.77 \times 10^{-8} \text{ cm} + 0.46 \times 10^{-8} \text{ cm}) = 1.42 \times 10^{-8} \text{ cm}$$

–8 –8 3 3 7. 2-20 The compound aluminum phosphide (AlP) is a compound semiconductor having mixed ionic and

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